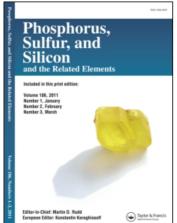
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## Phosphorus, Sulfur, and Silicon and the Related Elements

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# Syntheses, Structures and Photoelectron Spectra of Phospha-Alkenes and Phospha-Alkynes and Their Transition Metal Complexes

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SYNTHESES, STRUCTURES AND PHOTOELECTRON SPECTRA OF PHOSPHA-ALKENES AND PHOSPHA-ALKYNES AND THEIR TRANSITION METAL COMPLEXES

J.C.T.R. BURCKETT ST. LAURENT, P.B. HITCHCOCK, M.A. KING, H.W. KROTO, M.F. MEIDINE, S.I. KLEIN, S.I. AL-RESAYES, R.J. SUFFOLK and J.F. NIXON
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The chemistry of the novel phospha-alkenes RP=CR'<sub>2</sub>, and phospha-alkynes, RC=P, containing  $2p_{\pi}-3p_{\pi}$  bonds is of current interest. Recent molecular orbital calculations, 3,4 suggest that the highest occupied molecular orbital in CH<sub>2</sub>=PH is of the  $\pi$ -type with the phosphorus lone pair  $\sigma$ -orbital only slightly more stable while the  $\pi^*$  lumo orbital is relatively low lying. We now report He(I) photoelectron spectroscopic studies on a variety of RC=P molecules ,6 which indicate that the homo is also of the  $\pi$ -type and the  $\pi$ - $\sigma$  separation is much greater than that found in the analogous RC=N systems.

It might therefore be expected that phospha-alkenes are likely to act as both  $\eta^1-$  (phosphorus donors) and  $\eta^2-$  (P=C  $\pi\text{-donors})$  as in (a) and (b) below

whereas in suitable transition metal complexes the phospha-alkynes might be envisaged to act as either 2e, 4e or 6e donors by making use of the PEC  $\pi$ -system and the lone pair on phosphorus (c)  $\rightarrow$  (e). Recently we reported the first examples of complexes in which the

phospha-alkyne,  $^{t}$ BuC=P acts as a 2e donor in  $[Pt(PPh_{3})_{2}(P=CBu^{t})]$  as in (c) and as a 4e donor in  $[Co_{2}(CO)_{6}(P=CBu^{t})]$  as in (d).

We now describe the syntheses and first fully structurally characterised complexes in which the  ${}^{\rm t}$ BuC $^{\rm E}$ P acts as a 6e donor.

Treatment of  $^{t}BuC\equiv P$  with  $[Co_{2}(CO)_{8}]$  in toluene readily affords the deep red, light- and air-sensitive liquid complex  $[Co_{2}(CO)_{6}(P\equiv CBu^{t})]$  (I), which reacts with  $[W(CO)_{5}THF]$  to give light- and air-stable, wine-red crystals of the complex  $[Co_{2}(CO)_{6}(\mu^{-t}BuCP)W(CO)_{5}]$  (II). Similarly  $^{t}BuC\equiv P$  reacts with  $[(n^{5}-c_{5}H_{5})(CO)_{2}Mo\equiv Mo(CO)_{2}(n^{5}-c_{5}H_{5})]$  followed by  $[W(CO)_{5}(THF)]$  to give  $[n^{5}-(c_{5}H_{5})_{2}Mo_{2}(CO)_{4}(\mu^{-t}BuCP)W(CO)_{5}]$  (III).

$$(CO)_{3}CO \xrightarrow{PW(CO)_{5}} CO(CO)_{3} CO(CO)_{2}MO \xrightarrow{\text{PW}(CO)_{5}} CD(CO)_{2}CD$$

Single crystal X-ray structure determinations on (II) and (III) have been carried out and the molecular parameters will be discussed in relation to other complexes of this structural type, e.g.  $[(\text{Co}(\text{CO})_3)_2\text{E}_2]$  (E = P,As) and  $[(\text{Mo}(\text{C}_5\text{H}_5)(\text{CO})_2)_4\text{S}_2]$ . Of particular interest is the establishment of the side-on coordination of the phospha-alkyne to the  $\text{Co}_2$  or  $\text{Mo}_2$  system and the attachment of the W(CO) $_5$  unit via the phosphorus lone pair. The C-P bond length is much longer than that found for the free phospha-alkyne but interestingly is closer to that expected for a phospha-alkene than for a formal P-C single bond.

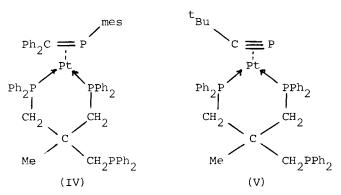
The volatility of  $[{\rm Co}_2({\rm CO})_6^{\rm t}{\rm BuCP}]$ , (I) offers a unique opportunity to compare its He(I) photo-electron spectrum with both that of the free phospha-alkyne and the alkyne complexes  $[{\rm Co}_2({\rm CO})_6({\rm RCCR})]$ . The ready interconversion of the dinuclear

complex  $[Pt(PPh_3)_2(PC^tBu)]$  to the dinuclear complex  $[(Pt(PPh_3)_2)_2PC^tBu]$  will also be described.

Previously we and others established the type (a)  $n^1$ -bonding for the phospha-alkene P(mesityl)=CPh<sub>2</sub> in single crystal X-ray studies on cis-PtCl<sub>2</sub>(PEt<sub>3</sub>)(P(mesityl)=CPh<sub>2</sub>), Cr(CO)<sub>5</sub>-(P(mesityl)=CPh<sub>2</sub>)) and Pt(PPh<sub>3</sub>)<sub>2</sub>P(mesityl)=CPh<sub>2</sub>) 10. The latter complex showed a solution nmr spectrum also supporting the possible existence of an  $n^2$ -isomer.

We now describe  $^{11}$  the synthesis of an  $^{2}$ -phospha-alkene complex Pt(triphos) (P(mesity1)=CPh $_{2}$ ) (IV) (two isomers) (triphos =  $(PPh_{2}CH_{2})_{3}CMe$ ) formed unexpectedly in the displacement reaction of PPh $_{3}$  from Pt(PPh $_{3}$ ) (triphos). The  $^{2}$ -phospha-alkyne complex Pt(triphos) (PEC $^{t}$ Bu), (V) is formed in a similar way.  $^{31}$ P and  $^{195}$ Pt nmr spectra will be presented and discussed.

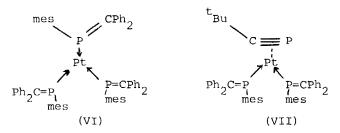
The  $\eta^2$ -mode of coordination of both (IV) and (V) is unambiguously established by the unusually small values of  $^1J_{\text{Pt-P}}$  for the coordinated phospha-alkene and phospha-alkyne which reflect the large s-character of the phosphorus lone pair



of the phospha-alkene and phospha-alkyne which is directed away from the metal atom. A similar effect was noted by us in the  $^2$ -Pt(PPh $_3$ ) $_2$ (P $\equiv$ C $^t$ Bu) complex.

In contrast to the above, treatment of  $Pt(cod)_2$ , (cod = 1,5-cyclooctadiene), with  $P(mesity1)=CPh_2$  gave the  $\eta^1$ -complex  $Pt(P(mesity1)=CPh_2)_3$  (VI), likewise when a 2:1 mixture of

P(mesity1)=CPh<sub>2</sub> and <sup>t</sup>BuCP was reacted with Pt(cod)<sub>2</sub> the product is Pt(P(mesity1)=CPh<sub>2</sub>)<sub>2</sub>(PEC<sup>t</sup>Bu) (VII) which is an interesting example of a complex containing only ligated phospha-alkenes ( $\eta^{1}$ -) and a phospha-alkyne ( $\eta^{2}$ -). Clearly a delicate balance of factors can result in either  $\eta^{1}$ - or  $\eta^{2}$ -coordination of the phospha-alkene.



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